AMENDMENTS TO THE SPECIFICATION

Replace the first full paragraph on page 10 with the following:

Further, isotopic concentrations provide stable isotopic identifications which are highly

specific. Elements which have more than one stable isotope are numerous. Of the 83 known

non-radioactive elements known to exist on earth, 62 have more than one stable isotope, and 40

have more than two stable isotopes. The element tin (Sn) has the largest number of stable

isotopes for any single element. Among the 40 elements having more than two stable isotopes,

there are a total of 224 stable isotopes. Although a few of the 224 252 stable isotopes are slightly

radioactive, they have very long lives and are present in many naturally occurning elements.

Thus, as will be seen, the stable isotopic identifications of the invention are numerous and

provide a ready and available means by which any product (including all pharmaceutical phases

APIs, drug products, excipients of drug products and/or impurities of drug products) may be

readily identified.

Replace the first full paragraph on page 13 with the following:

Additionally, the "error of identification" can be reduced or, the "precision of

identification" can be increased by choosing more than one isotopic concentration. There are a

total of 13 if one limits the stable isotopic identification of the invention to the common light

elements. Reduced error can be accomplished by using any number of the total of 252 available

stable isotopes elements having two or more stable isotopes.

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